Sc9 Name:

**COVALENT COMPOUNDS – PRACTICE**

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**Write the names of the following compounds**

|  |  |
| --- | --- |
| 1. N2O
 | f. N2O2 |
| 1. CO
 | g. P4S10 |
| 1. PI3
 | h. S2F10 |
| 1. PCl5
 | i. NI3 |
| 1. SO2
 | j. NO |

**Write the formulas of the following compounds**

|  |  |
| --- | --- |
| 1. Nitrogen tribromide
 | f. sulphur trioxide |
| 1. Sulphur hexafluoride
 | g. phosphorus pentabromide |
| 1. Dinitrogen tetrasulphide
 | h. diiodine hexachloride |
| 1. Oxygen difluoride
 | i. dichlorine monoxide |
| 1. Carbon tetraiodide
 | j. xenon hexafluoride |