Sc9 Name:

**COVALENT COMPOUNDS – PRACTICE**

**A picture containing indoor, table, computer, sitting

Description automatically generated**

**Write the names of the following compounds**

|  |  |
| --- | --- |
| 1. N2O | f. N2O2 |
| 1. CO | g. P4S10 |
| 1. PI3 | h. S2F10 |
| 1. PCl5 | i. NI3 |
| 1. SO2 | j. NO |

**Write the formulas of the following compounds**

|  |  |
| --- | --- |
| 1. Nitrogen tribromide | f. sulphur trioxide |
| 1. Sulphur hexafluoride | g. phosphorus pentabromide |
| 1. Dinitrogen tetrasulphide | h. diiodine hexachloride |
| 1. Oxygen difluoride | i. dichlorine monoxide |
| 1. Carbon tetraiodide | j. xenon hexafluoride |